

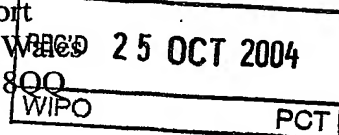


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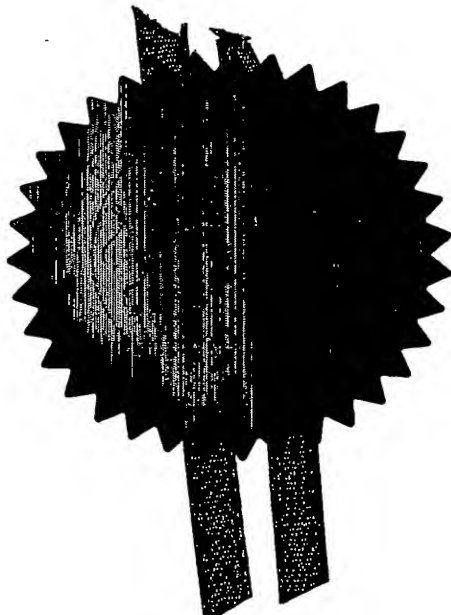
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11SEP03 E86665-2 D10002
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2. Patent application number (The Patent Office will fill in this part)	11 SEP 2003		0321287.5
3. Full name, address and postcode of the or of each applicant (underline all surnames)	Safeglass (Europe) Limited Whitworth Building Scottish Enterprise Technology Park East Kilbride GLASGOW G75 0QD		
08712010001 Patents ADP number (if you know it)			
If the applicant is a corporate body, give the country/state of its incorporation	UK		
4. Title of the invention	Glass like material with improved safety characteristics		
5. Name of your agent (if you have one)	Kennedys Patent Agency Limited		
"Address for service" in the United Kingdom to which all correspondence should be sent (including the postcode)	Floor 9, Queens House 29 St Vincent Place Glasgow G1 2DT		
Patents ADP number (if you know it)	0805 824 0002		
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8. Is a statement of inventorship and of right to grant of a patent required in support of this request? (Answer 'Yes' if:	Yes		
a) any applicant named in part 3 is not an inventor, or b) there is an inventor who is not named as an applicant, or c) any named applicant is a corporate body. See note (d))			

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Description

13 ✓

Claim(s)

Abstract

Drawing(s)

1 only

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Statement of inventorship and right to grant of a patent (Patents Form 7/77)

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11.

I/We request the grant of a patent on the basis of this application.

Signature
KENNEDYSDate
11 September 2003

12. Name and daytime telephone number of person to contact in the United Kingdom

Karen Veitch

Tel: 0141 226 6826

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Patents Form 1/77

1 Glass like material with improved safety characteristics.

2

3 The present invention relates to a glass-like material
4 which has improved safety characteristics compared to
5 ordinary glass.

6

7 In the present Application references to a "glass-like"
8 material refer to a material having the following
9 characteristics; clarity, brittleness, low strain to
10 failure and rigidity.

11

12 Ordinary glass is used in a variety of everyday
13 applications. For example it is known in the art to use
14 glass as a protective covering over fire and other types
15 of emergency alarms, emergency door releases, emergency
16 stop buttons on public transport, fire extinguishers,
17 fire axes and the like. As glass is transparent persons
18 can quickly and easily identify the presence of the alarm
19 or apparatus in the retaining box. If required, the
20 glass can be broken in order to access the alarm or
21 device.

22

23 However an inherent problem lies in the manner in which
24 the glass can be broken. Often a subsidiary device such

1 as a hammer is supplied with, or near to, the alarm or
2 apparatus, and can be used to break the glass. However,
3 in the event that this device is missing or cannot be
4 located in an emergency situation it will be necessary
5 for the person who wishes to access the apparatus or
6 alarm to break the glass by some other means. In the
7 event of an emergency situation the person may use, for
8 example, a hand or elbow for this purpose, and may, as a
9 result, sustain injuries from breaking the glass.

10 Breakage of glass results in the production of sharp
11 glass fragments and splinters, which can cause injury to
12 the user or other persons in the proximity of the alarm
13 or apparatus. In addition the potential risk of injury
14 from breaking the glass may cause hesitation on the part
15 of the person who wishes to access the alarm or
16 apparatus, having dangerous consequences.

17
18 The glass may also be broken by malicious or accidental
19 damage. Whilst the glass fragments can be removed and
20 the retaining glass replaced, there is an interim risk of
21 injury to persons coming into contact with the broken
22 fragments.

23
24 Considerable research has been conducted to find
25 materials which can be used in Applications similar to
26 glass but which minimise the risk of damage to persons in
27 the instance of the material being broken either
28 intentionally or accidentally. Safety glass i.e.
29 toughened glass, materials are well known in the art and
30 have numerous applications and uses. Most have enhanced
31 safety by virtue of being reinforced in strength, such
32 that they have a higher stress to failure than glass; in
33 other words a greater force is needed to shatter or break
34 them than would be required with ordinary glass. Uses

1 vary from windows and doors on automobiles and public
2 transport, to domestic uses such as shower enclosures and
3 room partitions. Whilst in many instances these have
4 greatly increased safety, they are of limited use in
5 Applications where it is actually desirable for the glass
6 to be broken, i.e. when used on retaining boxes of
7 emergency apparatus and alarms, because of their enhanced
8 strength and resistance to force.

9
10 For example the polymeric materials PerspexTM and
11 PlexiglasTM, are transparent like glass, and do not pose
12 the same risk of injury when broken. However, these
13 materials can be harder to break than glass and can still
14 produce sharp fragments when broken.

15
16 It is therefore an object of the present invention to
17 provide a material which resembles glass, but which has
18 improved safety characteristics when compared to ordinary
19 glass.

20
21 According to a first aspect of the present invention
22 there is provided a material which shatters, when broken,
23 into fragments which do not cut, puncture or otherwise
24 damage human skin or tissue, wherein the material is
25 comprised of an amorphous thermoplastic polymer and one
26 or more low molecular weight resins.

27
28 Preferably the material is comprised of a simple mixture
29 of amorphous thermoplastic polymer and one or more low
30 molecular weight resins.

31
32 Preferably the amorphous thermoplastic polymer is chosen
33 from the group consisting of polystyrene (PS), polymethyl
34 methacrylate (PMAA), styrene-acrylonitrile copolymer

1 (SAN), linear polyesters and co-polyesters and
2 polycarbonate (PC).

3

4 The one or more low molecular weight resins chosen will
5 be completely compatible with the chosen polymer. For
6 example in the case of polystyrene the low molecular
7 weight resin is typically C9 aromatic hydrocarbon resin.

8

9 Preferably the material has a tensile stress limit of
10 between 11 and 60 Nmm^{-2} .

11

12 Preferably the low molecular weight resin will have a Mn
13 (number average molecular weight) such that it has less
14 than 500 repeating units, and preferably less than 50
15 repeating units.

16

17 The material may be manufactured in sheet form.

18

19 According to a second aspect of the present invention
20 there is provided a polymeric blend comprising a polymer
21 selected from the group consisting of: polystyrene, (PS),
22 polymethyl methacrylate (PMAA), styrene-acrylonitrile
23 copolymer (SAN), linear polyesters and co-polyesters and
24 polycarbonate (PC) and one or more low molecular weight
25 resins.

26

27 The one or more low molecular weight resins chosen will
28 be completely compatible with the chosen polymer. For
29 example in the case of polystyrene the low molecular
30 weight resin is typically C9 aromatic hydrocarbon resin.

31

32 Preferably the one or more low molecular weight resins
33 have a Mn (number average molecular weight) such that it

1 has less than 500 repeating units, and preferably less
2 than 50 repeating units.

3

4 Preferably the one or more low molecular weight resins
5 are hydrocarbon resins.

6

7 Preferably the one or more low molecular weight resins
8 are aromatic hydrocarbon resins.

9

10 The polymeric blend may be manufactured in sheet form.

11

12 According to a third aspect of the present invention
13 there is provided a material which shatters, when broken,
14 into fragments which do not cut, puncture or damage human
15 skin or tissue, the material being comprised of

16 polystyrene and one or more low molecular weight resins.

17

18 Preferably the material is comprised of a simple mixture
19 of polystyrene and one or more low molecular weight
20 resins.

21

22 Preferably the one or more low molecular weight resins
23 are hydrocarbon resins.

24

25 Preferably the one or more low molecular weight resins
26 are aromatic hydrocarbon resins.

27

28 Most preferably the one or more low molecular weight
29 hydrocarbon resins are C9 aromatic hydrocarbon resins.

30

31 Preferably the one or more low molecular weight resins
32 are, or are derived from, alpha methyl styrene.

33

1 Preferably the one or more low molecular weight
2 hydrocarbon resins are selected from a group consisting
3 of; NorsoleneTM, KristalexTM, PlastolynTM EndexTM,
4 PiccotexTM, PiccolasticTM, SukorezTM or ArkonTM.

5

6 Most preferably the one or more low molecular weight
7 hydrocarbon resins are selected from a group consisting
8 of; Norsolene W90TM, Norsolene W100TM, Norsolene W110TM,
9 Kristalex F85TM, Kristalex F100TM, Kristalex F115TM,
10 Plastolyn 240TM, Plastolyn 290TM, Endex 155TM,
11 Piccolastic D125TM, Sukorez 100TM, Sukorez 120TM, Arkon
12 P100TM, Arkon P125TM, Arkon P140TM, Piccotex 75TM,
13 Piccotex 100TM or Piccotex 120TM.

14

15 Preferably the one or more low molecular weight resins
16 will have a Mn (number average molecular weight) such
17 that it has less than 500 repeating units, and preferably
18 less than 50 repeating units.

19

20 Preferably the material has a tensile stress limit
21 between 11 and 60 Nmm⁻².

22

23 Optionally the material may also include UV inhibitors,
24 antioxidants, flow modifiers, fire retarding agents,
25 colour pigments and brighteners as known in the art.

26

27 The material may be manufactured in sheet form.

28

29 According to a fourth aspect of the present invention
30 there is provided a method of manufacturing a material
31 which shatters, when broken, into which do not cut,
32 puncture or damage human skin or tissue, the method
33 comprising the step of mixing an amorphous thermoplastic
34 polymer and one or more low molecular weight resins.

1

2 Preferably the amorphous thermoplastic polymer is chosen
3 from the group consisting of polystyrene (PS), polymethyl
4 methacrylate (PMAA), styrene-acrylonitrile copolymer
5 (SAN), linear polyesters and co-polyesters and
6 polycarbonate (PC).

7

8 Preferably the one or more low molecular weight resins
9 are completely compatible with the chosen polymer. For
10 example in the case of polystyrene preferably the chosen
11 low molecular weight resin is C9 aromatic hydrocarbon
12 resin.

13

14 Preferably the one or more low molecular weight resins
15 are hydrocarbon resins.

16

17 Preferably the one or more low molecular weight resins
18 are aromatic hydrocarbon resins.

19

20 Preferably the low molecular weight resin will have a Mn
21 (number average molecular weight) such that it has less
22 than 500 repeating units, and preferably less than 50
23 repeating units.

24

25 Preferably as the polystyrene is mixed with the one or
26 more low molecular weight hydrocarbon resins, the glass
27 transition temperature (T_g) of the material is elevated.
28 Typically the T_g is elevated to 5-10 degrees C higher
29 than the base polymer.

30

31 According to a fifth aspect of the present invention
32 there is provided a method of manufacturing a material
33 which shatters, when broken, into fragments which do not
34 cut, puncture or damage human skin or tissue, the method

1 comprising the step of mixing polystyrene and one or more
2 low molecular weight hydrocarbon resins.

3

4 Preferably the one or more low molecular weight resins
5 are hydrocarbon resins.

6

7 Preferably the one or more low molecular weight resins
8 are aromatic hydrocarbon resins.

9

10 Most preferably the one or more low molecular weight
11 hydrocarbon resins are C9 aromatic hydrocarbon resins.

12

13 Preferably the one or more low molecular weight resins
14 are, or are derived from, alpha methyl styrene.

15

16 Preferably the one or more low molecular weight

17 hydrocarbon resins are selected from a group consisting of:

18 of; NorsoleneTM, KristalexTM, PlastolynTM, EndexTM,

19 PiccotexTM, PiccolasticTM, SukorezTM or ArkonTM.

20

21 Most preferably the one or more low molecular weight
22 hydrocarbon resins are selected from a group consisting
23 of; Norsolene W90TM, Norsolene W100TM, Norsolene W110TM,
24 Kristalex F85TM, Kristalex F100TM, Kristalex F115TM,
25 Plastolyn 240TM, Plastolyn 290TM, Endex 155TM,
26 Piccolastic D125TM, Sukorez 100TM, Sukorez 120TM, Arkon
27 P100TM, Arkon P125TM, Arkon P140TM, Piccotex 75TM,
28 Piccotex 100TM or Piccotex 120TM.

29

30 Preferably the low molecular weight resin will have a Mn
31 (number average molecular weight) such that it has less
32 than 500 repeating units, and preferably less than 50
33 repeating units.

34

1 The method may comprise the optional step of adding an
2 additive selected from the group consisting of UV
3 inhibitors, antioxidants, flow modifiers, fire retarding
4 agents, colour pigments and brighteners as known in the
5 art.

6

7 Preferably as the polystyrene is mixed with the one or
8 more low molecular weight hydrocarbon resins, the glass
9 transition temperature (T_g) of the material is elevated.
10 Typically the T_g is elevated to 5-10 degrees C higher
11 than the base polymer.

12

13 The material herein described can be used as a substitute
14 for ordinary glass. The material is glass-like in
15 character having clarity, brittleness, low strain to
16 failure and rigidity. The material has a variety of uses
17 including applications as enclosures and boxes to house
18 emergency equipment e.g. keys, first aid boxes, fire
19 extinguisher, window hammers, emergency stop buttons,
20 emergency kick out panels and alarms, as well as use in
21 access panels, windows and doors. It should be
22 recognised that the abovedescribed uses are by way of
23 example only and are not intended to limit the manner in
24 which the material is used. The material can be
25 manufactured in sheet form, by extrusion, and moulded
26 into any shape by injection moulding or other standard
27 melt processes.

28

29 Table 1 shows the stress-strain behaviour of the material
30 in comparison to other polystyrene materials. Figure 1
31 shows this information in the form of a graph.

32

Table : Comparison of Properties of Safeglass™ to Polystyrenes.

Modulus	Yield	Yield	Stress	Strain	Vicat
---------	-------	-------	--------	--------	-------

Polymer Type:	/ GPa	stress / MPa	strain / %	at break / MPa	at break / %	softening temperature
Polystyrene (i.e. "crystal" or GPPS)	3.0 - 3.2	Brittle - no yield		4 - 75	2	82 - 98
Toughened polystyrene (e.g. HIPS)	1.6 - 2.4	18 - 38	1.8	< yield	15 - >50	76 - 95
Safeglass™	3.1 - 3.4	Brittle - no yield		8 - 40	1 - 2	95 - 104

1
2 N.B. Safeglass™ materials are slightly more rigid and
3 certainly more brittle than conventional
4 "crystal" polystyrene. Modified polystyrenes are
5 invariably less rigid and tougher materials as a result
6 of blending with a rubbery (low T_g) additive. This also
7 results in a lowering of the Glass Transition Temperature
8 (T_g) as witnessed by the reduction in the Vicat Softening
9 Temperature. The reverse is true of Safeglass™ materials
10 which show no such decrease in T_g ; indeed it can be
11 higher than c.

12
13 The material is fundamentally a blend of a rigid and
14 normally brittle amorphous thermoplastic with a glass
15 transition temperature T_g at least 5° C above ambient and
16 one or more compatible low molecular weight resins.

17
18 An example embodiment will now be described by way of
19 example only.

20
21 A rigid and normally brittle amorphous thermoplastic
22 polymer is blended with one or more low molecular weight
23 resins which have a M_n (number average molecular weight)
24 such that the resin has less than 500 repeating units,
25 preferably less than 50 repeating units. The one or more
26 low molecular weight resins have a weight average
27 molecular weight of 6050 or below.

28

1 The material is manufactured by mixing or blending a
2 clear polymer with one or more low molecular weight
3 hydrocarbon resins. The polymer is an amorphous
4 thermoplastic and can be chosen from the group of
5 polystyrene, (PS), polymethyl methacrylate (PMAA),
6 styrene-acrylonitrile copolymer (SAN), linear polyesters
7 and co-polyesters and polycarbonate (PC). It is important
8 that the low molecular weight resin is completely
9 compatible with the chosen polymer. For example in the
10 case of polystyrene it is C9 aromatic hydrocarbon resin.

11

12 In the herein described embodiment polystyrene is used.

13

14 The one or more low molecular weight resins which are
15 mixed with the polystyrene are aromatic hydrocarbon
16 resins and typically C9 aromatic hydrocarbon resins.
17 The one or more resins are typically alpha-methyl styrene
18 or vinyl toluene or derivatives thereof. These are
19 selected from the following group: Norsolene W90TM,
20 Norsolene W100TM, Norsolene W110TM, Kristalex F85TM,
21 Kristalex F100TM, Kristalex F115TM, Plastolyn 240TM,
22 Plastolyn 290TM, Endex 155TM, Piccolastic D125TM, Sukorez
23 100TM, Sukorez 120TM, Arkon P100TM, Arkon P125TM, Arkon
24 P140TM, Piccotex 75TM, Piccotex 100TM or Piccotex 120TM.

25

26 It has been discovered that by blending polystyrene with
27 one or more of the abovementioned low molecular weight
28 hydrocarbon resins, a hard, rigid material is formed
29 which has the appearance and feel of glass, but which is
30 extremely brittle and has low strain to failure. The
31 material also has the inherent advantage that when
32 broken, unlike glass, the material breaks into fragments
33 which are not sharp and do not injure skin or tissue. The
34 material is, by design, manufactured to break between 11

1 and 60 Nmm⁻². Therefore the material, when provided as a
2 substitute to glass, for example in retaining boxes for
3 emergency devices and alarms, can easily be broken by a
4 human hand, fist, elbow, foot or the like and
5 advantageously shatters into fragments or pieces which
6 are not sharp and are not capable of cutting or
7 puncturing human skin. Due to the inherent advantages of
8 the material it is envisaged that it may have a variety
9 of other uses, for example it may have application in
10 novelty toys, such as stress relief toys, or have uses in
11 "stunt" apparatus in, for example, theatres, shows or on
12 film sets.

13

14 The material is manufactured by conventional melt
15 compounding techniques. As the polystyrene is mixed with
16 the one or more low molecular weight hydrocarbon resins,
17 the glass transition temperature (T_g) of the material is
18 elevated as the low molecular weight resin does not have
19 a plasticising effect, the opposite effect is seen as the
20 glass transition temperature of the material is elevated.

21

22 The material is generally transparent or clear, however
23 dyes may be added to change the appearance of the
24 material.

25

26 Low molecular weight in resins is a function of the
27 length of the chains in the resin. In this case the
28 hydrocarbon resins have a very low molecular weight, too
29 low in fact for the resins to be of any use on their own,
30 and are difficult to mould. By mixing low molecular
31 weight hydrocarbon resin with polystyrene, the stress
32 limit of the polystyrene is reduced giving the material
33 the characteristics described in the present Application.
34 Preferably the low molecular weight resin will have a Mn

1 (number average molecular weight) such that it has less
2 than 500 repeating units, and preferably less than 50
3 repeating units.

4

5 The following is an example of the material of the
6 present invention.

7

8 **Example 1**

9 In order to achieve a material with a stress limit of 24
10 Mpa, a 50% mix of polymer and 50% resin is used, which
11 achieves this stress limit. Typically the polymer could
12 be crystal polystyrene such as Polystyrol™ 143E, and
13 resin Plastolyn™ 240.

14

15 **Example 2**

16 In order to achieve a material with a stress limit of 34
17 Mpa, a 60% mix of polymer and 40% resin is used, which
18 achieves this stress limit. Typically the polymer could
19 be crystal polystyrene such as Polystyrol™ 143E, and
20 resin Plastolyn™ 240.

21

22 Further modifications and improvements may be added
23 without departing from the scope of the invention herein
24 intended.

25

1/1

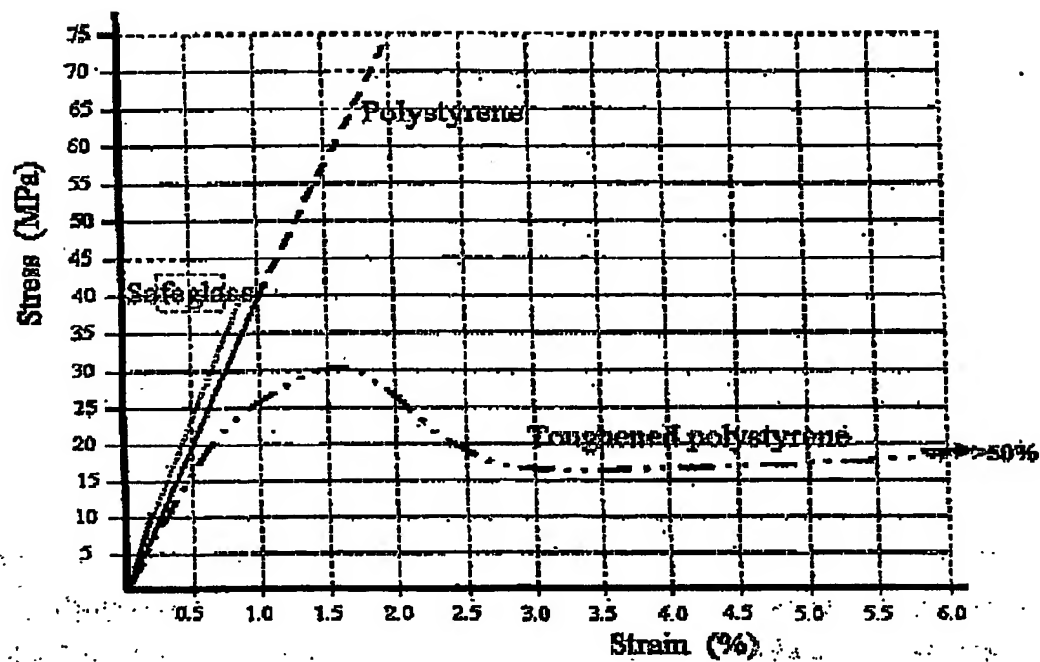


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